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to decrease heat gradually from a house at a rate of  $900 \text{ kJ/min}$  while drawing electric power at a rate of  $75 \text{ kW}$ . Determine the COP of this system and the rate of heat transfer to the outside weather.

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The Clausius-Clapeyron equation:  $dP_i^{\text{sat}} = \frac{h_i^{\text{vap}}}{RT^2} dT$  or  $\ln \frac{P_i^{\text{sat}}}{101 \text{ kPa}} = \frac{h_i^{\text{vap}}}{R} \left( \frac{1}{1373 \text{ [K]}} - \frac{1}{T} \right)$  so  $P_i^{\text{sat}} = 101 \text{ kPa} \exp \left( \frac{h_i^{\text{vap}}}{R} \left( \frac{1}{1373 \text{ [K]}} - \frac{1}{T} \right) \right)$

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Chapter 6 2 If the fluid is in thermodynamic equilibrium any thermodynamic variable for a pure substance, like pure water, can be written in terms of any two other thermodynamic variables, i.e.  $p=p(\rho,T)$  (6.1.1) where the functional relationship in depends on the substance.



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