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Naming Ions & Counting Atoms: Chapter 9.1 Naming & Writing Formulas for Ionic Compounds: Chapter 9.2 Naming & Writing Formulas for Molecular Compounds: Chapter 9.3 Naming & Writing Formulas for Acids: Chapter 9.4 Quizlet Chapter 9 vocabulary Answer Keys Ionic Compound Naming and Formula Examples from notes

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Section 9.4 - Naming and Writing Formulas for Acids and Bases. An acid is a compound that produces H^+ ions when it dissolves in water. The formula for an acid normally starts with an H. When naming acids, you should first determine whether or not the acid contains oxygen.

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Chapter 9 Chemical Names and Formulas 83 SECTION 9.3 NAMING AND WRITING FORMULAS FOR MOLECULAR COMPOUNDS (pages 268–270) This section explains the rules for naming and writing formulas for binary molecular compounds. Naming Binary Molecular Compounds (pages 268–269)

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a. 2- b. 1+ c. 1- d. 3+ 43. a. 2+ b. 2+ c. 3+ d.1+ 44. a. barium ion b. iodide ion c. silver ion d. mercury(II) ion 45. cyanide, CN- and hydroxide, OH- 46. a. hydroxide ion b. lead(IV) ion c. sulfate ion d. oxide ion 47. zero; A compound is electrically neutral. 48. The symbols for the cation and

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Chapter 9 Chemical Names and Formulas 221 Name Date 4

Write formulas for the following ionic compounds a sodium

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phosphate c sodium hydroxide e ammonium chloride b magnesium d potassium cyanide 5 Write formulas for compounds formed from these pairs of ions b , 6 Name the following compounds ...

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Chapter 9 "Chemical Names and Formulas" ... in parenthesis after the name of the metal (Table 9.2, p.255) Predicting Ionic Charges Some of the post-transition elements also have more than one possible oxidation state. Tin (II) = Sn^{2+} Lead (II) = Pb^{2+} Tin (IV) = Sn^{4+} Lead (IV) = Pb^{4+}

"Chemical Names and Formulas"

Chapter 9 Chemical Names and Formulas 211. Section Review. Objectives. • Determine the charges of monatomic ions by using the periodic table and write the names of the ions. • Define a polyatomic ion and write the names and formulas of the most

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common polyatomic ions. •Identify the two common endings for the names of most polyatomic ions.

05 CTR ch09 7/9/04 3:29 PM Page 211 NAMING IONS 9

Chapter 9 Chemical Names and Formulas 221 Name Date 4

Write formulas for the following ionic compounds a sodium phosphate c sodium hydroxide e ammonium chloride b magnesium d potassium cyanide 5 Write formulas for compounds formed from these pairs of ions b , 6 Name the following compounds ...

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