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NO₃ (aq) 0 K (aq) NO₃ (aq) AgI(s)

Chemical Reactions

In 30 minutes, $5 \times 10^5 \text{ J s}^{-1} \text{ m}^{-2} \times 1800 \text{ s}$ gives $9 \times 10^8 \text{ J m}^{-2}$, which corresponds to 1.36×10^{27} molecules m^{-2} , or 1.36×10^{23} molecules cm^{-2} . Download CHAP9 gh rev 2 - Solution manual Physical Chemistry

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