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SWBAT

determine the mass of one substance involved in a chemical reaction if given the mass of any one of the other substances involved in the reaction. in other words . . .

Slide 1

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121 Section 8.1 The
Covalent Bond pages
240–247 Practice

Problems page 244

Draw the Lewis
structure for each
molecule. 1. PH_3 H_2O H_2S
 $\text{H}-\text{H}$ $\text{H}-\text{H}$ $\text{P}-\text{P}$ respectively,
for single, double, and
triple $\text{P}-\text{P}$ — — 2. H_2S H_2O
 $\text{H}-\text{H}$ — $\text{H}-\text{S}-\text{S}-\text{H}$ 3. HCl H_2O
 $\text{Cl}-\text{H}$ — $\text{Cl}-\text{H}$ 4. CCl_4 Cl_2 Cl_2
 $\text{Cl}-\text{C}-\text{C}-\text{Cl}$ — —

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Assessment page 372

5. Compare the mass of the reactants and the mass of the products in a chemical reaction, and explain how these masses are related. The coefficients in the balanced equation indicate the molar relationship between each ...

Stoichiometry

Mass Finals for CHEM

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metric prefix kilo
means: 1,000 1000
—442.70

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Experiment 1 Report

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DISCUSSION

QUESTIONS I. Explain
the following terms:
accuracy of data,
precision of data, and
instrumental precision.

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2 2. The density of a substance is considered an inherent property, unique to that substance. Is this property constant under all conditions? Explain. 3.

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Chemistry Section

Review 6.3 - 6.4

Multiple Choice Identify the choice that best completes the statement or answers the question. Put the LETTER of the correct

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answer in the blank.

Assessment

1. The chemical formula for an ionic compound represents the a. number of atoms in each molecule. b. number of ions in each molecule.

Chemistry Section Review 6.3 - 6

Chapter 12 (pages 320 - 321) - Answer questions - 1-9, 14.

End-of-Chemistry Assessment Review. To help you prepare for

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Assessment, review

the following at the
end of each chapter:

Chapter Highlights.

Vocabulary . Answer

Chapter Review

questions (answers will

be posted on my

website next week)

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