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CHAPTER 11 REVIEW
Gases Class SHORT ANSWER Answer the following questions in the space provided. c c
The molar mass of a gas at STP is the density of that gas (a)

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multiplied by the mass
of 1 mol. (b) divided by
the mass of 1 mol. nRT
(c) multiplied by 22.4
L. (d) divided by 22.4 L.
For the expression $V =$
(a) increasing P (b)
decreasing T

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Chapter 11 Chemical Reactions Test Answer Key

Chapter 11 Gases 459
t's Monday morning,
and Lilia is walking out
of the chemistry
building, thinking
about the introductory

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lecture on gases that her instructor just presented. Dr. Scanlon challenged the class to try to visualize gases in terms of the model she described, so Lilia looks at her hand and tries to picture the particles in the air

**Chapter 11 Gases -
An Introduction to
Chemistry**

BJU Chemistry Chapter
11 Vocabulary.

Intermolecular force.

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Dipole-dipole force.

Hydrogen bond.

Dispersion force

(London force) an electrostatic attraction between molecules; it is much weaker.... the attraction of the positive end of one polar molecule to the negative end of another.... a special dipole-dipole force involving hydrogen and a highly electronegative atom....

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chapter 11

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11.1 Word Equations

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Questions 1. What is the law of conservation of mass?

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See Figure 11.1 for a
sketch of Rutherford's
atom. Rutherford could

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not answer questions such as “Why aren’t the negative electrons attracted into the positive nucleus, causing the

Chapter 11 Modern Atomic Theory

Describe two ways that atmospheric nitrogen is changed to form useful compounds.

nitrogen—fixing bacteria, Haber process to make NH_3

Chapter 11 Typical

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Elements 137 Chapter
Review 25. 26. 138 In
1669, Hemiig Brandt
noticed a new
substance was
produced when he
heated urine and sand
together in an attempt
to make gold.

Chapter 11 Review

Answers -

TYPICAL, \ufb02

ELEMENTS 11.1 ...

Chapter 1 Review

Solutions: 11 The mass

has four significant

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figures, while the density has three significant figures, so the answer must have three significant figures. 22. It is silver. To determine whether or not it is silver, we can calculate its density. However, the

Answer Key and Tests for Discovering Design with Sample ...

AP Chemistry Chapter
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Multiple-choice
exercise. Choose the
correct answer for each
question. Show all
questions $\leq \Rightarrow$ How
many milliliters of 6.0
M HNO_3 are needed to
prepare 500 mL of 0.50
M HNO_3 ? ? 0.25 mL ?
40 mL ? 15.76 mL ?
300 mL ? None of
these are correct ...

AP Chemistry
Chapter 11 Review
Questions

Chapter 11 Modern
Page 22/26

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Answers
Atomic Theory. Review
Skills 11.1 The
Mysterious Electron
Standing Waves and
Guitar Strings Electrons
as Standing Waves
Waveforms for
Hydrogen Atoms
Particle Interpretation
of the Wave Character
of the Electron Other
Important Waveforms
Overall Organization of
Principal Energy
Levels, Sublevels, and
Orbitals 11.2 Multi-
Electron Atoms.

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Atomic Theory - An
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CHAPTER 11 REVIEW
Gases SECTION 2
SHORT ANSWER

Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease,

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or stay the same in the following circumstances:

- a. temperature increases, volume stays the same
- b. volume increases, temperature stays the same

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