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chemist wants to
separate a mixture of
salt and fat. He places
the mixture in a

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container with the nonpolar solvent benzene (density = 0.8787 g/cm^3) and water (density = 1.00 g/cm^3). After shaking the solution and allowing it to settle, the correct arrangement of solvent and solute is:

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4.3. For a reaction, $A + B \rightarrow \text{Product}$; the rate law is given by, $r = k [A]^{1/2} [B]^2$. What is the order of the reaction? Ans. Order of reaction. $= 1/2 + 2 = 2.5$ or 2.5.

4.4. The conversion of molecules X to Y follows second order kinetics. If concentration of X is

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increased to three
times how will it affect
the rate of formation of
Y ?

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Smallest partical of an element that retains its identity in a. . .

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Answer: According to VSEPR theory, if CH_4 were square planar, the bond angle would be 90° . For tetrahedral structure, the bond angle is $109^\circ 28'$. For tetrahedral structure, the bond angle is $109^\circ 28'$.

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the following
compounds in
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