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Summary Questions AQA GCSE

Chemistry C14 The Earth's Resources

Kerboodle Answers: Page No. 218. a.

Answer. The boiling point reduces by
reducing the pressure. So to reduce the

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boiling point, the pressure is reduced so that less energy is required in boiling.

1. Answer.

New(9-1) AQA GCSE Chemistry C14 The Earth's Resources ...

C 12.2 Analysing chromatograms AQA
GCSE Chemistry C12 Chemical Analysis
Kerboodle Answers: Page No. 183

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Answer- Paper chromatography is a method for separating dissolved substances from one another. It is often used when the dissolved substances are colored, such as inks, food coloring and plant dyes.

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Chemical Analysis ...**

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C9.2 Fractional distillation of oil AQA
GCSE Chemistry C9 Hydrocarbon
Kerboodle Answers Page No 151. a. The
boiling point of hydrocarbons increases
as the number of carbon atoms in the

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molecule increases. This is because the forces between molecules increase as the size of the molecule increases.

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Chemistry C6 Electrolysis Kerboodle

Answers: Page No. 110 1. Positive ions

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move towards Cathode (-) Oxidation
happens at Anode (+)

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answers : Secondary ...

C8.3 The effect of Temperature AQA
GCSE Chemistry C8 Rates and
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- 133 1.a Answer Increasing the
temperature increases reaction rates
because of the increase in the number of
high energy collisions. Increasing the
temperature makes the particles to

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collide more often.

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lesson of chemistry to show. Please have Qs ready to ask during consolidation lessons in September.

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Revision

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Kerboodle Resources Answers? - The Student Room

$\text{NaCl(s)} + \text{H}_2\text{SO}_4(\text{aq}) \rightarrow \text{NaHSO}_4(\text{aq}) + \text{HCl(g)}$ Sodium bromide first of all reacts with concentrated sulfuric acid to make hydrogen bromide. $\text{NaBr(s)} + \text{H}_2\text{SO}_4(\text{aq}) \rightarrow \text{NaHSO}_4(\text{aq}) + \text{HBr(g)}$ However, the bromide ions produced are strong enough reducing agents to reduce the

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sulfuric acid which is present to sulphur dioxide.

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